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(9*H*-Carbazol-9-ylmethyl)diethylamine

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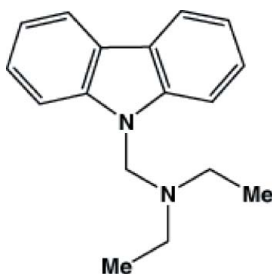
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Key indicators: single-crystal X-ray study; $T = 291$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.048; wR factor = 0.120; data-to-parameter ratio = 15.7.

The asymmetric unit of the title compound, $\text{C}_{17}\text{H}_{20}\text{N}_2$, contains two molecules, whose bond lengths and angles differ only slightly. In the crystal, neighbouring molecules form pillar structures *via* edge-to-face π - π stacking interactions [edge-to-face distances = 3.538 (3) and 3.496 (3) Å].

Related literature

Carbazole-based compounds are widely used in OLEDs as emitters because of their intense luminescence, see: Adhikari *et al.* (2007); Liu *et al.* (2006); Palayangoda *et al.* (2008) and as organic fluorescence probes, see: Hao *et al.* (2010); Pappayee & Mishra, (2000). For our studies of organic fluorescence probes, see: Shen *et al.* (2006, 2008). For the preparation of the title compound, see: Gu *et al.* (1997).



Experimental

Crystal data

 $\text{C}_{17}\text{H}_{20}\text{N}_2$ $M_r = 252.35$

Monoclinic, $P2_1/c$
 $a = 24.338$ (2) Å
 $b = 6.3216$ (11) Å
 $c = 19.133$ (2) Å
 $\beta = 104.109$ (2)°
 $V = 2854.9$ (6) Å³

$Z = 8$
 Mo $K\alpha$ radiation
 $\mu = 0.07$ mm⁻¹
 $T = 291$ K
 $0.28 \times 0.24 \times 0.22$ mm

Data collection

Bruker SMART APEX CCD diffractometer
 Absorption correction: multi-scan (*SADABS*; Bruker, 2000)
 $T_{\min} = 0.981$, $T_{\max} = 0.985$

13429 measured reflections
 5463 independent reflections
 3052 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.048$
 $wR(F^2) = 0.120$
 $S = 1.06$
 5463 reflections

347 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.16$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.18$ e Å⁻³

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINTE* (Bruker, 2000); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG2745).

References

- Adhikari, R. M., Mondal, R., Shah, B. K. & Neckers, D. C. (2007). *J. Org. Chem.* **72**, 4727–4732.
 Bruker (2000). *SMART*, *SAINTE* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Gu, W., Pan, Y., Shen, Y., Huang, X., Sun, H. & Sun, X. (1997). *Acta Cryst.* **C53**, 190–191.
 Hao, L.-L., Li, J.-F., Lin, P.-H., Dong, R.-F., Li, D.-X., Shuang, S.-M. & Dong, C. (2010). *Chin. Chem. Lett.* **21**, 9–12.
 Liu, Y., Nishiura, M., Wang, Y. & Hou, Z. (2006). *J. Am. Chem. Soc.* **128**, 5592–5593.
 Palayangoda, S. S., Cai, X., Adhikari, R. M. & Neckers, D. C. (2008). *Org. Lett.* **10**, 281–284.
 Pappayee, N. & Mishra, A. K. (2000). *Spectrochim. Acta Part A*, **56**, 1027–1034.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Shen, Y.-M., Wang, B.-X., Feng, Y.-Y., Shen, Z.-Y., Shen, J., Li, C. & Hu, H.-W. (2006). *Chem. J. Chin. Univ.* **27**, 651–653.
 Shen, Z.-Y., Wang, B.-X., Shen, J. & Hu, H.-W. (2008). *Chem. J. Chin. Univ.* **29**, 916–918.

supplementary materials

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(9*H*-Carbazol-9-ylmethyl)diethylamine

W.-J. Gu, Z.-T. Lin and B.-X. Wang

Comment

Carbazole-based compounds are known for their intense luminescence and widely used in OLEDs as emitters (Adhikari *et al.*, 2007; Liu *et al.*, 2006; Palayangoda *et al.*, 2008). They can also be used as organic fluorescence probes (Hao *et al.*, 2010; Pappayee *et al.*, 2000). In our continuing studies in organic fluorescence probes (Shen *et al.*, 2008; Shen *et al.*, 2006), the 9-diethylaminomethyl carbazole(I) was synthesized.

The crystal structure of the title compound, C₁₇H₂₀N₂, reveals that all the bond lengths and angles have normal values. Each asymmetric unit contains two title molecules, which are similar to each other with only slightly difference in their bond lengths and angles (Figure 1 and Table 1).

In the crystal packing the edge-to-face π - π stacking interactions were observed. The distance from the edge of the molecular plane B (N1ⁱ/C1ⁱ/C2ⁱ/C3ⁱ/C4ⁱ/C5ⁱ/C6ⁱ/C7ⁱ/C8ⁱ/C9ⁱ/ C10ⁱ/C11ⁱ/C12ⁱ)(i: 1 - x, -1/2 + y, 0.5 - z) to the face of the molecular plane A (N1/C1/C2/C3/C4/C5/C6/C7/C8/C9/C10/C11/C12) is 3.538 (3) Å, and the dihedral angle between plane A and B is 62.06 (3)°. Similar relationships were observed with the molecular plane C and D. The edge-to-face distance of the molecular plane D (N3ⁱⁱ/C18ⁱⁱ/C19ⁱⁱ/C20ⁱⁱ/C21ⁱⁱ/C22ⁱⁱ/C23ⁱⁱ/ C24ⁱⁱ/C25ⁱⁱ/C26ⁱⁱ/C27ⁱⁱ/C28ⁱⁱ/C29ⁱⁱ)(ii: -x, 1/2 + y, 0.5 - z) to the molecular plane C (N3/C18/C19/C20/C21/C22/C23/C24/C25/C26/C27 /C28/C29) is 3.496 (3) Å, and the dihedral angle between plane C and D is 61.41 (3)° (Figure 2). Through these edge-to-face π - π stacking interactions, the neighbouring molecules form pillar structures (Figure 3).

Experimental

9-Diethylaminomethyl carbazole was prepared according to a procedure described in the literature (Gu, *et al.*, 1997). Colourless crystals were obtained by recrystallized from ethanol at room temperature.

¹H-NMR (CDCl₃, 400 MHz) δ : 1.07 (t, 6H, 2-CH₃), 2.68 (q, 4H, 2-CH₂-), 4.98 (s, 2H, -CH₂-), 7.22 (t, 2H, ArH), 7.44 (t, 2H, ArH), 7.54 (d, 2H, ArH), 8.06 (d, 2H, ArH).

Refinement

The H atoms were placed in calculated positions and included as part of a riding model, with C—H = 0.93–0.97 Å, and with U_{equiv} values set at 1.2 U_{equiv} of the parent atoms.

Figures

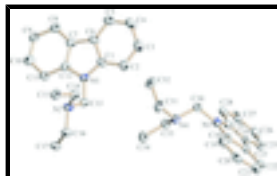


Fig. 1. A view of the title compound showing the atom-numbering scheme and displacement ellipsoids drawn at 30% probability level. All H atoms have been omitted for clarity.



Fig. 2. A view of the edge-to-face π - π stacking interactions. Dashed lines indicate weak edge-to-face π - π stacking interactions and all H atoms have been omitted for clarity. (i: $1 - x, -1/2 + y, 0.5 - z$ ii: $-x, 1/2 + y, 0.5 - z$)

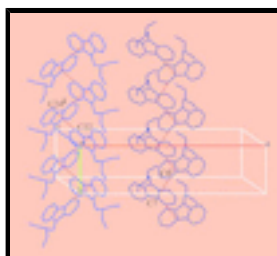


Fig. 3. A view of the pillar structure. Dashed lines indicate weak edge-to-face π - π stacking interactions. All H atoms have been omitted for clarity. (i: $1 - x, -1/2 + y, 0.5 - z$ ii: $-x, 1/2 + y, 0.5 - z$)

(9H-Carbazol-9-ylmethyl)diethylamine

Crystal data

$C_{17}H_{20}N_2$

$M_r = 252.35$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 24.338$ (2) Å

$b = 6.3216$ (11) Å

$c = 19.133$ (2) Å

$\beta = 104.109$ (2)°

$V = 2854.9$ (6) Å³

$Z = 8$

$F(000) = 1088$

$D_x = 1.174$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 1470 reflections

$\theta = 2.2$ – 21.0 °

$\mu = 0.07$ mm⁻¹

$T = 291$ K

Block, colourless

$0.28 \times 0.24 \times 0.22$ mm

Data collection

Bruker SMART APEX CCD diffractometer

Radiation source: sealed tube graphite

phi and ω scans

Absorption correction: multi-scan (SADABS; Bruker, 2000)

$T_{\min} = 0.981$, $T_{\max} = 0.985$

13429 measured reflections

5463 independent reflections

3052 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.039$

$\theta_{\text{max}} = 26.0$ °, $\theta_{\text{min}} = 0.9$ °

$h = -25 \rightarrow 29$

$k = -7 \rightarrow 7$

$l = -22 \rightarrow 18$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.048$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.120$	H-atom parameters constrained
$S = 1.06$	$w = 1/[\sigma^2(F_o^2) + (0.05P)^2]$
5463 reflections	where $P = (F_o^2 + 2F_c^2)/3$
347 parameters	$(\Delta/\sigma)_{\max} < 0.001$
0 restraints	$\Delta\rho_{\max} = 0.16 \text{ e } \text{\AA}^{-3}$
	$\Delta\rho_{\min} = -0.18 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Least-squares planes (x,y,z in crystal coordinates) and deviations from them (* indicates atom used to define plane)

$$8.9285(0.0063)x - 3.2585(0.0028)y + 12.6587(0.0061)z = 2.4983(0.0046)$$

* -0.0251 (0.0014) N1 * -0.0167 (0.0017) C1 * 0.0499 (0.0017) C2 * 0.0607 (0.0018) C3 * -0.0014 (0.0018) C4 * -0.0354 (0.0017) C5 * -0.0486 (0.0017) C6 * -0.0407 (0.0017) C7 * 0.0096 (0.0017) C8 * 0.0577 (0.0017) C9 * 0.0343 (0.0017) C10 * -0.0110 (0.0016) C11 * -0.0333 (0.0017) C12

Rms deviation of fitted atoms = 0.0374

$$8.9285(0.0063)x + 3.2585(0.0028)y + 12.6588(0.0061)z = 11.1303(0.0032)$$

Angle to previous plane (with approximate e.s.d.) = 62.06 (0.03)

* 0.0251 (0.0014) N1_\$1 * 0.0167 (0.0017) C1_\$1 * -0.0499 (0.0017) C2_\$1 * -0.0607 (0.0018) C3_\$1 * 0.0014 (0.0018) C4_\$1 * 0.0354 (0.0017) C5_\$1 * 0.0486 (0.0017) C6_\$1 * 0.0407 (0.0017) C7_\$1 * -0.0096 (0.0017) C8_\$1 * -0.0577 (0.0017) C9_\$1 * -0.0343 (0.0017) C10_\$1 * 0.0110 (0.0016) C11_\$1 * 0.0333 (0.0017) C12_\$1

Rms deviation of fitted atoms = 0.0374

$$-17.1148(0.0065)x + 3.2280(0.0027)y + 12.4596(0.0056)z = 1.3402(0.0013)$$

Angle to previous plane (with approximate e.s.d.) = 67.15 (0.03)

* -0.0095 (0.0014) N3 * -0.0154 (0.0016) C18 * -0.0081 (0.0017) C19 * 0.0048 (0.0018) C20 * 0.0356 (0.0017) C21 * 0.0196 (0.0016) C22 * -0.0218 (0.0017) C23 * -0.0287 (0.0017) C24 * -0.0272 (0.0016) C25 * -0.0051 (0.0017) C26 * 0.0333 (0.0018) C27 * 0.0347 (0.0016) C28 * -0.0123 (0.0016) C29

Rms deviation of fitted atoms = 0.0225

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$$-17.1148(0.0065)x - 3.2280(0.0027)y + 12.4596(0.0056)z = 3.2755(0.0028)$$

Angle to previous plane (with approximate e.s.d.) = 61.41 (0.03)

* 0.0095 (0.0014) N3_\$2 * 0.0154 (0.0016) C18_\$2 * 0.0081 (0.0017) C19_\$2 * -0.0048 (0.0018) C20_\$2 * -0.0356 (0.0017) C21_\$2 * -0.0196 (0.0016) C22_\$2 * 0.0218 (0.0017) C23_\$2 * 0.0287 (0.0017) C24_\$2 * 0.0272 (0.0016) C25_\$2 * 0.0051 (0.0017) C26_\$2 * -0.0333 (0.0018) C27_\$2 * -0.0347 (0.0016) C28_\$2 * 0.0123 (0.0016) C29_\$2

Rms deviation of fitted atoms = 0.0225

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.36426 (8)	0.9208 (3)	0.17614 (10)	0.0376 (4)
C2	0.31632 (9)	0.8391 (4)	0.19419 (11)	0.0458 (5)
H2	0.3006	0.7112	0.1752	0.055*
C3	0.29289 (10)	0.9544 (4)	0.24125 (11)	0.0539 (6)
H3	0.2610	0.9027	0.2543	0.065*
C4	0.31618 (11)	1.1462 (4)	0.26928 (12)	0.0580 (6)
H4	0.2990	1.2226	0.2996	0.070*
C5	0.36412 (10)	1.2249 (3)	0.25303 (10)	0.0469 (5)
H5	0.3799	1.3513	0.2733	0.056*
C6	0.38897 (8)	1.1136 (3)	0.20581 (10)	0.0395 (5)
C7	0.43876 (9)	1.1426 (3)	0.17880 (10)	0.0380 (5)
C8	0.48144 (10)	1.2920 (3)	0.19114 (11)	0.0440 (5)
H8	0.4804	1.4063	0.2214	0.053*
C9	0.52563 (9)	1.2705 (3)	0.15822 (12)	0.0489 (6)
H9	0.5547	1.3698	0.1671	0.059*
C10	0.52730 (9)	1.1005 (3)	0.11143 (12)	0.0471 (5)
H10	0.5570	1.0900	0.0889	0.057*
C11	0.48519 (9)	0.9490 (3)	0.09856 (10)	0.0413 (5)
H11	0.4860	0.8360	0.0677	0.050*
C12	0.44188 (8)	0.9704 (3)	0.13283 (10)	0.0363 (4)
C13	0.38250 (9)	0.6526 (3)	0.08628 (10)	0.0390 (5)
H13A	0.3519	0.5759	0.0993	0.047*
H13B	0.4153	0.5601	0.0946	0.047*
C14	0.31317 (9)	0.8340 (3)	-0.00586 (11)	0.0422 (5)
H14A	0.3163	0.9432	0.0305	0.051*
H14B	0.2817	0.7431	-0.0028	0.051*
C15	0.30047 (10)	0.9368 (3)	-0.07941 (11)	0.0494 (5)
H15A	0.3329	1.0161	-0.0846	0.074*
H15B	0.2686	1.0299	-0.0844	0.074*
H15C	0.2919	0.8295	-0.1160	0.074*
C16	0.35944 (9)	0.5149 (3)	-0.03411 (11)	0.0446 (5)

H16A	0.3432	0.5523	-0.0841	0.054*
H16B	0.3331	0.4205	-0.0190	0.054*
C17	0.41427 (9)	0.3987 (3)	-0.02917 (12)	0.0468 (5)
H17A	0.4436	0.4985	-0.0311	0.070*
H17B	0.4098	0.3016	-0.0687	0.070*
H17C	0.4244	0.3220	0.0154	0.070*
C18	0.04584 (8)	0.0516 (3)	0.15594 (10)	0.0351 (4)
C19	0.00476 (9)	0.0810 (4)	0.09248 (11)	0.0446 (5)
H19	0.0056	0.1976	0.0631	0.053*
C20	-0.03736 (9)	-0.0685 (4)	0.07439 (12)	0.0520 (6)
H20	-0.0650	-0.0531	0.0315	0.062*
C21	-0.04000 (9)	-0.2432 (4)	0.11849 (12)	0.0522 (6)
H21	-0.0696	-0.3397	0.1053	0.063*
C22	0.00128 (8)	-0.2722 (3)	0.18142 (11)	0.0436 (5)
H22	-0.0001	-0.3881	0.2109	0.052*
C23	0.04508 (8)	-0.1255 (3)	0.20027 (10)	0.0364 (4)
C24	0.09355 (8)	-0.1064 (3)	0.26133 (10)	0.0365 (4)
C25	0.11486 (9)	-0.2287 (4)	0.32240 (11)	0.0446 (5)
H25	0.0971	-0.3544	0.3295	0.054*
C26	0.16223 (9)	-0.1619 (4)	0.37195 (11)	0.0503 (6)
H26	0.1768	-0.2443	0.4125	0.060*
C27	0.18903 (9)	0.0279 (4)	0.36267 (11)	0.0494 (5)
H27	0.2209	0.0708	0.3973	0.059*
C28	0.16878 (9)	0.1523 (3)	0.30273 (11)	0.0431 (5)
H28	0.1865	0.2791	0.2968	0.052*
C29	0.12153 (8)	0.0847 (3)	0.25158 (10)	0.0343 (4)
C30	0.10458 (9)	0.3792 (3)	0.15881 (11)	0.0409 (5)
H30A	0.1289	0.4608	0.1971	0.049*
H30B	0.0694	0.4569	0.1422	0.049*
C31	0.12229 (10)	0.5486 (4)	0.05375 (12)	0.0543 (6)
H31A	0.0819	0.5757	0.0397	0.065*
H31B	0.1345	0.5186	0.0101	0.065*
C32	0.15171 (11)	0.7484 (3)	0.08706 (14)	0.0623 (7)
H32A	0.1410	0.7785	0.1311	0.093*
H32B	0.1408	0.8643	0.0542	0.093*
H32C	0.1920	0.7292	0.0970	0.093*
C33	0.19210 (9)	0.3007 (3)	0.12612 (12)	0.0464 (5)
H33A	0.1946	0.1725	0.1546	0.056*
H33B	0.2113	0.4122	0.1576	0.056*
C34	0.22266 (11)	0.2653 (4)	0.06688 (14)	0.0658 (7)
H34A	0.2016	0.1667	0.0324	0.099*
H34B	0.2598	0.2094	0.0874	0.099*
H34C	0.2259	0.3972	0.0434	0.099*
N1	0.39612 (7)	0.8375 (2)	0.13156 (8)	0.0365 (4)
N2	0.36559 (6)	0.7080 (2)	0.00993 (8)	0.0359 (4)
N3	0.09245 (7)	0.1786 (2)	0.18752 (8)	0.0353 (4)
N4	0.13199 (7)	0.3584 (2)	0.09955 (9)	0.0391 (4)

supplementary materials

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0431 (11)	0.0382 (10)	0.0286 (9)	0.0088 (9)	0.0031 (8)	0.0040 (8)
C2	0.0447 (12)	0.0515 (13)	0.0404 (11)	0.0040 (10)	0.0085 (10)	0.0047 (10)
C3	0.0502 (13)	0.0698 (16)	0.0412 (12)	0.0061 (12)	0.0100 (11)	0.0061 (11)
C4	0.0629 (15)	0.0736 (17)	0.0377 (12)	0.0200 (14)	0.0129 (11)	-0.0067 (12)
C5	0.0649 (15)	0.0367 (11)	0.0328 (10)	0.0090 (10)	-0.0003 (11)	-0.0009 (9)
C6	0.0433 (11)	0.0376 (11)	0.0311 (9)	0.0080 (9)	-0.0037 (9)	0.0007 (8)
C7	0.0425 (11)	0.0332 (10)	0.0303 (10)	0.0068 (9)	-0.0062 (8)	0.0003 (8)
C8	0.0568 (13)	0.0342 (10)	0.0335 (10)	0.0041 (10)	-0.0032 (10)	-0.0009 (8)
C9	0.0454 (12)	0.0396 (11)	0.0527 (13)	-0.0095 (10)	-0.0056 (11)	0.0034 (10)
C10	0.0382 (11)	0.0453 (12)	0.0540 (13)	-0.0005 (10)	0.0036 (10)	-0.0018 (10)
C11	0.0440 (11)	0.0342 (11)	0.0423 (11)	0.0037 (9)	0.0038 (10)	-0.0023 (9)
C12	0.0346 (10)	0.0307 (10)	0.0384 (10)	0.0082 (8)	-0.0011 (8)	0.0021 (8)
C13	0.0434 (11)	0.0300 (9)	0.0415 (11)	0.0006 (9)	0.0059 (9)	-0.0054 (9)
C14	0.0420 (11)	0.0409 (11)	0.0405 (11)	0.0065 (9)	0.0039 (10)	0.0005 (9)
C15	0.0531 (13)	0.0404 (11)	0.0490 (12)	-0.0012 (10)	0.0015 (10)	0.0074 (9)
C16	0.0457 (12)	0.0346 (10)	0.0504 (11)	0.0004 (9)	0.0057 (10)	-0.0151 (9)
C17	0.0482 (13)	0.0360 (11)	0.0533 (13)	0.0020 (9)	0.0071 (10)	-0.0125 (9)
C18	0.0368 (10)	0.0385 (10)	0.0329 (9)	0.0014 (8)	0.0144 (9)	-0.0014 (8)
C19	0.0453 (12)	0.0541 (12)	0.0358 (10)	0.0013 (11)	0.0126 (9)	-0.0022 (9)
C20	0.0386 (12)	0.0752 (16)	0.0411 (11)	0.0002 (11)	0.0074 (10)	-0.0129 (11)
C21	0.0415 (12)	0.0636 (15)	0.0562 (13)	-0.0151 (11)	0.0209 (11)	-0.0211 (12)
C22	0.0447 (12)	0.0429 (11)	0.0499 (12)	-0.0078 (10)	0.0246 (11)	-0.0083 (9)
C23	0.0372 (10)	0.0412 (11)	0.0350 (10)	0.0002 (9)	0.0168 (9)	-0.0025 (9)
C24	0.0407 (11)	0.0371 (10)	0.0360 (10)	0.0031 (9)	0.0175 (9)	-0.0021 (9)
C25	0.0480 (13)	0.0478 (12)	0.0436 (11)	0.0107 (10)	0.0216 (11)	0.0070 (10)
C26	0.0487 (13)	0.0693 (15)	0.0370 (12)	0.0180 (12)	0.0180 (10)	0.0081 (10)
C27	0.0374 (11)	0.0752 (16)	0.0353 (10)	0.0007 (11)	0.0080 (9)	-0.0005 (11)
C28	0.0456 (12)	0.0472 (12)	0.0405 (11)	-0.0020 (10)	0.0180 (10)	-0.0077 (9)
C29	0.0363 (10)	0.0382 (10)	0.0314 (9)	0.0026 (8)	0.0139 (8)	0.0008 (8)
C30	0.0459 (11)	0.0278 (10)	0.0530 (12)	0.0014 (9)	0.0194 (10)	-0.0005 (9)
C31	0.0524 (13)	0.0558 (14)	0.0514 (13)	-0.0038 (11)	0.0060 (11)	0.0236 (11)
C32	0.0694 (17)	0.0346 (12)	0.0866 (18)	0.0006 (11)	0.0263 (14)	0.0193 (12)
C33	0.0524 (13)	0.0381 (11)	0.0531 (13)	0.0045 (10)	0.0214 (11)	0.0052 (9)
C34	0.0804 (19)	0.0502 (14)	0.0831 (18)	0.0030 (13)	0.0516 (16)	0.0091 (13)
N1	0.0406 (9)	0.0327 (8)	0.0347 (9)	0.0007 (7)	0.0065 (7)	-0.0038 (7)
N2	0.0356 (8)	0.0309 (8)	0.0367 (9)	0.0040 (7)	0.0003 (7)	-0.0064 (7)
N3	0.0428 (9)	0.0309 (8)	0.0344 (8)	-0.0010 (7)	0.0135 (7)	0.0030 (6)
N4	0.0481 (10)	0.0327 (9)	0.0389 (9)	0.0014 (7)	0.0151 (8)	0.0065 (7)

Geometric parameters (\AA , $^\circ$)

C1—N1	1.389 (2)	C18—C19	1.384 (3)
C1—C2	1.394 (3)	C18—N3	1.401 (2)
C1—C6	1.415 (3)	C18—C23	1.408 (3)
C2—C3	1.385 (3)	C19—C20	1.375 (3)

C2—H2	0.9300	C19—H19	0.9300
C3—C4	1.389 (3)	C20—C21	1.401 (3)
C3—H3	0.9300	C20—H20	0.9300
C4—C5	1.372 (3)	C21—C22	1.380 (3)
C4—H4	0.9300	C21—H21	0.9300
C5—C6	1.394 (3)	C22—C23	1.392 (3)
C5—H5	0.9300	C22—H22	0.9300
C6—C7	1.440 (3)	C23—C24	1.449 (3)
C7—C8	1.381 (3)	C24—C25	1.392 (3)
C7—C12	1.413 (3)	C24—C29	1.421 (3)
C8—C9	1.379 (3)	C25—C26	1.369 (3)
C8—H8	0.9300	C25—H25	0.9300
C9—C10	1.406 (3)	C26—C27	1.398 (3)
C9—H9	0.9300	C26—H26	0.9300
C10—C11	1.380 (3)	C27—C28	1.379 (3)
C10—H10	0.9300	C27—H27	0.9300
C11—C12	1.377 (3)	C28—C29	1.384 (3)
C11—H11	0.9300	C28—H28	0.9300
C12—N1	1.390 (2)	C29—N3	1.390 (2)
C13—N1	1.445 (2)	C30—N3	1.440 (2)
C13—N2	1.460 (2)	C30—N4	1.455 (2)
C13—H13A	0.9700	C30—H30A	0.9700
C13—H13B	0.9700	C30—H30B	0.9700
C14—N2	1.471 (2)	C31—N4	1.473 (3)
C14—C15	1.512 (3)	C31—C32	1.514 (3)
C14—H14A	0.9700	C31—H31A	0.9700
C14—H14B	0.9700	C31—H31B	0.9700
C15—H15A	0.9600	C32—H32A	0.9600
C15—H15B	0.9600	C32—H32B	0.9600
C15—H15C	0.9600	C32—H32C	0.9600
C16—N2	1.470 (2)	C33—N4	1.473 (3)
C16—C17	1.506 (3)	C33—C34	1.516 (3)
C16—H16A	0.9700	C33—H33A	0.9700
C16—H16B	0.9700	C33—H33B	0.9700
C17—H17A	0.9600	C34—H34A	0.9600
C17—H17B	0.9600	C34—H34B	0.9600
C17—H17C	0.9600	C34—H34C	0.9600
N1—C1—C2	129.63 (19)	C19—C20—C21	122.1 (2)
N1—C1—C6	109.08 (18)	C19—C20—H20	119.0
C2—C1—C6	121.27 (19)	C21—C20—H20	119.0
C3—C2—C1	117.9 (2)	C22—C21—C20	120.0 (2)
C3—C2—H2	121.0	C22—C21—H21	120.0
C1—C2—H2	121.0	C20—C21—H21	120.0
C2—C3—C4	121.1 (2)	C21—C22—C23	119.1 (2)
C2—C3—H3	119.5	C21—C22—H22	120.4
C4—C3—H3	119.5	C23—C22—H22	120.4
C5—C4—C3	121.2 (2)	C22—C23—C18	119.69 (18)
C5—C4—H4	119.4	C22—C23—C24	133.32 (19)
C3—C4—H4	119.4	C18—C23—C24	106.96 (17)

supplementary materials

C4—C5—C6	119.4 (2)	C25—C24—C29	119.35 (19)
C4—C5—H5	120.3	C25—C24—C23	133.94 (19)
C6—C5—H5	120.3	C29—C24—C23	106.69 (17)
C5—C6—C1	119.1 (2)	C26—C25—C24	119.3 (2)
C5—C6—C7	134.7 (2)	C26—C25—H25	120.4
C1—C6—C7	106.15 (17)	C24—C25—H25	120.4
C8—C7—C12	118.8 (2)	C25—C26—C27	121.1 (2)
C8—C7—C6	133.57 (19)	C25—C26—H26	119.4
C12—C7—C6	107.61 (17)	C27—C26—H26	119.4
C9—C8—C7	119.48 (19)	C28—C27—C26	120.8 (2)
C9—C8—H8	120.3	C28—C27—H27	119.6
C7—C8—H8	120.3	C26—C27—H27	119.6
C8—C9—C10	120.9 (2)	C27—C28—C29	118.6 (2)
C8—C9—H9	119.6	C27—C28—H28	120.7
C10—C9—H9	119.6	C29—C28—H28	120.7
C11—C10—C9	120.6 (2)	C28—C29—N3	130.39 (18)
C11—C10—H10	119.7	C28—C29—C24	120.81 (18)
C9—C10—H10	119.7	N3—C29—C24	108.77 (16)
C12—C11—C10	117.93 (19)	N3—C30—N4	113.15 (15)
C12—C11—H11	121.0	N3—C30—H30A	108.9
C10—C11—H11	121.0	N4—C30—H30A	108.9
C11—C12—N1	129.44 (17)	N3—C30—H30B	108.9
C11—C12—C7	122.36 (19)	N4—C30—H30B	108.9
N1—C12—C7	108.18 (17)	H30A—C30—H30B	107.8
N1—C13—N2	111.93 (15)	N4—C31—C32	116.10 (18)
N1—C13—H13A	109.2	N4—C31—H31A	108.3
N2—C13—H13A	109.2	C32—C31—H31A	108.3
N1—C13—H13B	109.2	N4—C31—H31B	108.3
N2—C13—H13B	109.2	C32—C31—H31B	108.3
H13A—C13—H13B	107.9	H31A—C31—H31B	107.4
N2—C14—C15	113.21 (17)	C31—C32—H32A	109.5
N2—C14—H14A	108.9	C31—C32—H32B	109.5
C15—C14—H14A	108.9	H32A—C32—H32B	109.5
N2—C14—H14B	108.9	C31—C32—H32C	109.5
C15—C14—H14B	108.9	H32A—C32—H32C	109.5
H14A—C14—H14B	107.7	H32B—C32—H32C	109.5
C14—C15—H15A	109.5	N4—C33—C34	113.93 (19)
C14—C15—H15B	109.5	N4—C33—H33A	108.8
H15A—C15—H15B	109.5	C34—C33—H33A	108.8
C14—C15—H15C	109.5	N4—C33—H33B	108.8
H15A—C15—H15C	109.5	C34—C33—H33B	108.8
H15B—C15—H15C	109.5	H33A—C33—H33B	107.7
N2—C16—C17	113.83 (16)	C33—C34—H34A	109.5
N2—C16—H16A	108.8	C33—C34—H34B	109.5
C17—C16—H16A	108.8	H34A—C34—H34B	109.5
N2—C16—H16B	108.8	C33—C34—H34C	109.5
C17—C16—H16B	108.8	H34A—C34—H34C	109.5
H16A—C16—H16B	107.7	H34B—C34—H34C	109.5
C16—C17—H17A	109.5	C1—N1—C12	108.96 (15)

C16—C17—H17B	109.5	C1—N1—C13	126.47 (17)
H17A—C17—H17B	109.5	C12—N1—C13	124.41 (16)
C16—C17—H17C	109.5	C13—N2—C16	109.79 (15)
H17A—C17—H17C	109.5	C13—N2—C14	110.73 (15)
H17B—C17—H17C	109.5	C16—N2—C14	111.71 (15)
C19—C18—N3	129.58 (18)	C29—N3—C18	108.65 (15)
C19—C18—C23	121.49 (18)	C29—N3—C30	127.21 (17)
N3—C18—C23	108.92 (16)	C18—N3—C30	124.03 (16)
C20—C19—C18	117.6 (2)	C30—N4—C33	111.05 (16)
C20—C19—H19	121.2	C30—N4—C31	110.64 (16)
C18—C19—H19	121.2	C33—N4—C31	114.16 (17)
N1—C1—C2—C3	179.13 (19)	C23—C24—C25—C26	-178.5 (2)
C6—C1—C2—C3	1.1 (3)	C24—C25—C26—C27	0.8 (3)
C1—C2—C3—C4	0.4 (3)	C25—C26—C27—C28	-0.7 (3)
C2—C3—C4—C5	-2.0 (3)	C26—C27—C28—C29	-0.4 (3)
C3—C4—C5—C6	1.9 (3)	C27—C28—C29—N3	179.32 (19)
C4—C5—C6—C1	-0.4 (3)	C27—C28—C29—C24	1.5 (3)
C4—C5—C6—C7	-177.7 (2)	C25—C24—C29—C28	-1.4 (3)
N1—C1—C6—C5	-179.48 (17)	C23—C24—C29—C28	177.65 (17)
C2—C1—C6—C5	-1.1 (3)	C25—C24—C29—N3	-179.67 (16)
N1—C1—C6—C7	-1.5 (2)	C23—C24—C29—N3	-0.6 (2)
C2—C1—C6—C7	176.88 (18)	C2—C1—N1—C12	-176.97 (19)
C5—C6—C7—C8	1.5 (4)	C6—C1—N1—C12	1.2 (2)
C1—C6—C7—C8	-176.0 (2)	C2—C1—N1—C13	7.6 (3)
C5—C6—C7—C12	178.7 (2)	C6—C1—N1—C13	-174.21 (16)
C1—C6—C7—C12	1.2 (2)	C11—C12—N1—C1	178.27 (19)
C12—C7—C8—C9	0.4 (3)	C7—C12—N1—C1	-0.5 (2)
C6—C7—C8—C9	177.3 (2)	C11—C12—N1—C13	-6.2 (3)
C7—C8—C9—C10	1.0 (3)	C7—C12—N1—C13	175.11 (16)
C8—C9—C10—C11	-1.3 (3)	N2—C13—N1—C1	108.6 (2)
C9—C10—C11—C12	0.1 (3)	N2—C13—N1—C12	-66.2 (2)
C10—C11—C12—N1	-177.16 (18)	N1—C13—N2—C16	172.62 (17)
C10—C11—C12—C7	1.4 (3)	N1—C13—N2—C14	-63.6 (2)
C8—C7—C12—C11	-1.6 (3)	C17—C16—N2—C13	-66.3 (2)
C6—C7—C12—C11	-179.32 (17)	C17—C16—N2—C14	170.45 (18)
C8—C7—C12—N1	177.19 (17)	C15—C14—N2—C13	167.25 (16)
C6—C7—C12—N1	-0.5 (2)	C15—C14—N2—C16	-70.0 (2)
N3—C18—C19—C20	-178.99 (19)	C28—C29—N3—C18	-177.64 (19)
C23—C18—C19—C20	0.6 (3)	C24—C29—N3—C18	0.4 (2)
C18—C19—C20—C21	1.2 (3)	C28—C29—N3—C30	-1.4 (3)
C19—C20—C21—C22	-1.6 (3)	C24—C29—N3—C30	176.64 (16)
C20—C21—C22—C23	0.2 (3)	C19—C18—N3—C29	179.61 (19)
C21—C22—C23—C18	1.5 (3)	C23—C18—N3—C29	0.0 (2)
C21—C22—C23—C24	179.0 (2)	C19—C18—N3—C30	3.2 (3)
C19—C18—C23—C22	-1.9 (3)	C23—C18—N3—C30	-176.41 (16)
N3—C18—C23—C22	177.72 (16)	N4—C30—N3—C29	104.2 (2)
C19—C18—C23—C24	179.99 (17)	N4—C30—N3—C18	-80.1 (2)
N3—C18—C23—C24	-0.4 (2)	N3—C30—N4—C33	-75.7 (2)
C22—C23—C24—C25	1.7 (4)	N3—C30—N4—C31	156.47 (18)

supplementary materials

C18—C23—C24—C25	179.5 (2)	C34—C33—N4—C30	177.03 (18)
C22—C23—C24—C29	-177.1 (2)	C34—C33—N4—C31	-57.1 (2)
C18—C23—C24—C29	0.6 (2)	C32—C31—N4—C30	69.2 (3)
C29—C24—C25—C26	0.2 (3)	C32—C31—N4—C33	-56.9 (3)

Fig. 1

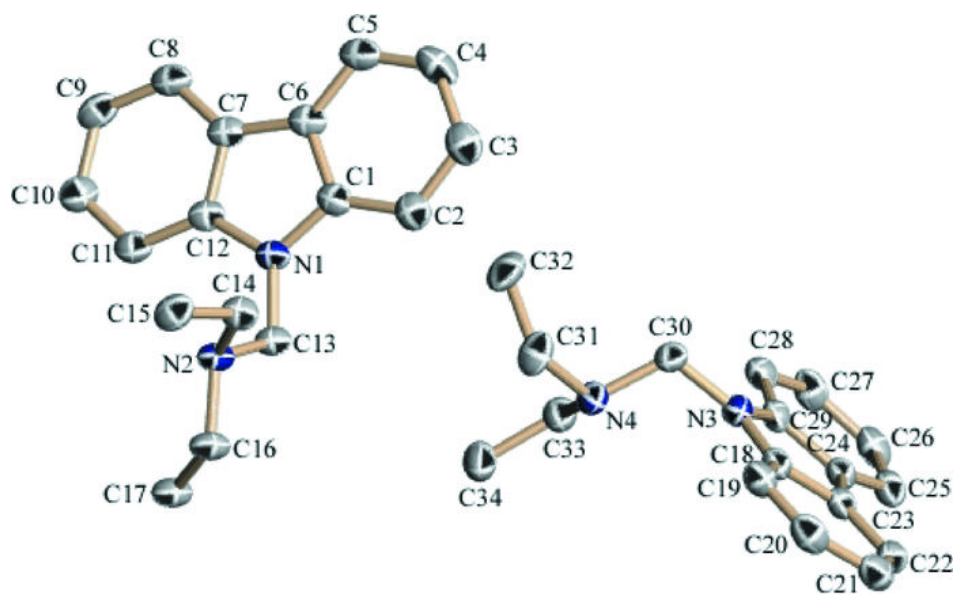


Fig. 2

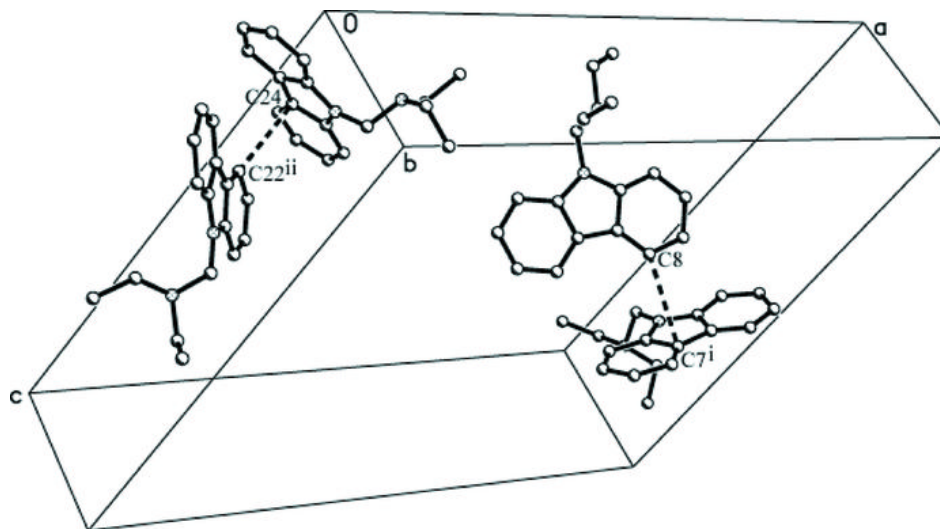


Fig. 3

